

The deuterium nucleus is compound of one proton and one neutron.

Because $M_p \approx M_n$ and $V_p \approx V_n$, we obtain:

$$V_{(p \text{ or } n)}^f = V_{(p \text{ or } n)}^i - J_{(p \text{ or } n)} M_{(n \text{ or } p)}$$

(see Vixra: 1711.0299. Author Piscedda Giampaolo),

then in the $Ox_1x_2x_3$, V_p or V_n vanishing, because its volume route of 90° almost completely, along the x_4 axis. Let $V_n \rightarrow 0$, so:

$$V_p^f = V_p^i - [V_p^i / (M_p + M_n)] M_\mu. \text{ From this formula we obtain } r_p^f = (\frac{3}{4\pi} V_p^f)^{1/3}.$$

Because the electron deuterium charge radius is $r_p^i + (r_d^i - r_p^i)$, then,

the muonic deuterium charge radius is: $r_d^i - r_p^i + r_p^f = r_d^f$.

$$r_p^i - r_p^i = 1,2673 \cdot 10^{-15} \text{ m}, V_p^i = 2,807 \cdot 10^{-45} \text{ m}^3, M_p + M_n = 3,347549 \cdot 10^{-27} \text{ kg}$$

$$V_p^i M_\mu / (M_p + M_n) = 1,579456 \cdot 10^{-46} \text{ m}^3; \text{ then } r_p^f = 8,58369 \cdot 10^{-16} \text{ m}. \text{ Then:}$$

$$r_d^i - r_p^i + r_p^f = r_d^f = 2,12567 \cdot 10^{-15} \text{ m}$$

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