

**TESTING OF THE
PROTOTYPE THERMAL
PROPERTIES (THP)
EXPERIMENT FOR THE
SURFACE SCIENCE
PACKAGE OF THE
CASSINI/HUYGENS PROBE**

1. ABSTRACT

The purpose of this project is to test the proposed circuitry and materials required in order to attempt to measure the thermal properties of the fluid (either atmosphere or ocean) expected to be found on the surface of Titan, a satellite of Saturn. The proposed experiment forms part of the contribution of the University of Kent at Canterbury Unit for Space Sciences to the Surface Science Package of the Huygens Probe, part of the joint ESA/NASA Cassini Mission to the Saturn system launched in 1996.

The experiment uses the “transient hot wire” method to obtain an accurate measure of both the thermal conductivity and diffusivity of a surrounding fluid. The advantage of this method is that it requires only a measurement of variation of resistance in a wire over a short period. This method is widely used in laboratory research, but has never before been attempted on such a small scale and at so great a distance.

2. INTRODUCTION

This project concerns the implementation of the “transient hot wire” method of thermal property measurement in a remote sensing device, part of the Surface Science Package on the Huygens Probe, part of the Cassini Mission to Titan, a satellite of Saturn, in 1996. This method, first described by Nagasaka and Nagashima¹ in 1981, has since become a common method of measurement of thermal properties on Earth. Put simply, the method involves passing a small (of the order of milliamps) current, in the form of a pulse or a constant amount, through a wire made of a pure conductor immersed in a fluid. For the purposes of this project, the constant current method was chosen for simplicity. The resulting change in potential difference between the ends of the wire (due to the change in resistance of the wire) bears a simple mathematical relationship to the length of time current has been flowing for the first few seconds after the current has begun to flow. This corresponds to the dissipation of heat from the wire by conduction. It is observed that after about ten seconds, loss of heat by convection in the fluid becomes a significant consideration, and the potential difference then deviates from the simple relationship mentioned above. It is therefore vital to record accurately and at high sampling frequency the changes in potential difference during the first few seconds of the experiment. The constant of proportionality in the voltage/time relationship (gradient of an appropriate graph) must be determined accurately in order to obtain accurate values of thermal conductivity and diffusivity using the equation derived in reference 1.

Previous work on this method by Grant² showed that a high sampling frequency and accuracy were necessary for this experiment and also that appropriate devices were not freely available. To this end, a custom-built circuit has now been provided, designed along the same lines as the circuit to be sent with the Huygens Probe. Carbon fibre has previously been used as the sensor material, but it is too fragile for a long mission with harsh landing conditions. Very thin (of the order of 50 μm diameter) platinum wire was therefore chosen, mainly because of its durability and high level of purity. Other methods of measurement were considered – for example the Needle Probe method used by Asher, Sloan and Grabowski⁹, but this method is not as accurate and is more difficult to construct.

The present project consisted of two parts:

- Establishment of the relationship between temperature and resistance for the wire, as this value is needed in the final set of equations (see Section 3).
- Writing of appropriate software for interfacing the analogue to digital converter to a PC and displaying the results, and obtaining measurements of potential difference at a sufficiently high sampling rate to yield accurate values for the thermal conductivity and diffusivity of water at room temperature.

3. THEORY

There are three main ways in which heat may be transferred through a fluid³:

i) Conduction

This is the mode of heat transfer due to intermolecular interactions in a fluid. Fourier first stated a law which fully describes the effect of this heat flow:

$$q = \frac{dQ}{dt} = -\lambda A \frac{d\Delta T}{dn} \quad (1)$$

where dQ/dt = heat flow in unit time
 λ = thermal conductivity
 A = cross-sectional area of fluid perpendicular to direction of heat flow
 $d\Delta T/dn$ = temperature gradient in direction normal to surface of the emitter

$$q'' = \frac{q}{A} \quad (2)$$

q'' is often termed the heat flow.

$$q'' = -\lambda \frac{d\Delta T}{dn} \quad (3)$$

is the most general form of Fourier's law.

ii) Radiation

This is the mode of heat transfer due to the emission of electromagnetic radiation from the emitting surface. It is described by the Stefan-Boltzmann law:

$$\frac{dQ}{dt} = \sigma A (T_1^4 - T_2^4) \quad (4)$$

where σ = Stefan-Boltzmann constant ($5.67 \times 10^{-8} \text{ Wm}^{-2}\text{K}^{-4}$)
 T_1, T_2 = temperatures of the two materials involved

iii) Convection

This mode of heat transfer results from the macroscopic movement of fluid around the emissive material and has the effect of altering the influence of one or both of the other modes of transfer. Two forms of convection occur:

Forced convection results from a mass movement of fluid brought about by the action of a pump or fan.

Natural or free convection results from a local reduction in the density of the fluid. This fluid of lower density will rise from the emitting surface, bringing about a transfer of heat independent of the other modes. It has been found that

$$\frac{dQ}{dt} = hA(T_1^4 - T_2^4) \quad (5)$$

where h is a constant of proportionality dependent on the nature of the fluid and emissive body
 A = area of emissive surface
 T_1, T_2 = temperatures of the emissive surface and the surrounding fluid respectively

As our aim is to determine the thermal conductivity and diffusivity of a fluid, the experiment must be designed to minimise the loss of heat by radiation and convection. From equation (4), it can be seen that heat flux due to radiation is dependent on the surface area of the emitting body. By employing a very thin (50 μm diameter) wire in this capacity, this mode of heat loss can be largely neglected. Convection, too, depends on the surface area of the body, but as will be seen (Section 3.1) the onset of convective losses is evident in the deviation of the voltage/ $\ln(\text{time})$ relationship from a straight line.

3.1 THEORY OF MEASUREMENTS

As

$$q_x'' = -\lambda \frac{\Delta T}{x} \quad (\text{from (2)})$$

Then in three dimensions

$$\underline{q}'' = -\lambda \text{grad} \Delta T \quad (6)$$

The heat flux, and therefore temperature, at any point in a material is a function of the three dimensions of space and time. Consider the case of a rectangular parallelepiped with a point P at its centre, its edges parallel to the x, y and z axes and of lengths $2dx$, $2dy$ and $2dz$ respectively. Call the faces in the planes $x-dx$ and $x+dx$ ABCD and A'B'C'D' respectively. Heat will flow across the face ABCD at a rate

$$\frac{dQ_{ABCD}}{dt} = 4 \left(q_x'' - \frac{\partial q_x''}{\partial x} dx \right) dydz \quad (7)$$

where q_x'' = flux at P across a parallel plane.

Similarly, the heat flow across A'B'C'D' is given by

$$\frac{dQ_{A'B'C'D'}}{dt} = 4 \left(q_x'' + \frac{\partial q_x''}{\partial x} dx \right) dydz \quad (8)$$

Therefore, the rate of gain of heat across both faces is given by

$$\frac{dQ_x}{dt} = -8 \left(\frac{\partial q_x''}{\partial x} \right) dx dy dz \quad (9)$$

Extending this result to three dimensions, the total rate at which heat is gained by the parallelepiped is

$$\frac{dQ_V}{dt} = -8 \left(\frac{\partial q_x''}{\partial x} + \frac{\partial q_y''}{\partial y} + \frac{\partial q_z''}{\partial z} \right) dx dy dz = -8 dx dy dz \operatorname{div} q_x'' \quad (10)$$

This is equivalent to

$$\frac{dQ_V}{dt} = 8\rho c \frac{\delta \Delta T}{\delta t} dx dy dz \quad (11)$$

where ρ = density
 c = specific heat capacity at temperature T

Equating (10) and (11) we have

$$\rho c \frac{\delta \Delta T}{\delta t} + \left(\frac{\partial q_x''}{\partial x} + \frac{\partial q_y''}{\partial y} + \frac{\partial q_z''}{\partial z} \right) = 0 \quad (12)$$

Equation (12) holds at all points in a material, as long as these points are not themselves sources of heat.

For a homogeneous isotropic solid with thermal conductivity which is independent of temperature, q_x'' , q_y'' and q_z'' are given by appropriate variations of equation (2). Equation (12) becomes

$$\frac{\delta^2 \Delta T}{\delta x^2} + \frac{\delta^2 \Delta T}{\delta y^2} + \frac{\delta^2 \Delta T}{\delta z^2} - \frac{1}{K} \frac{\delta \Delta T}{\delta t} = 0 \quad (13)$$

where

$$K = \text{thermal diffusivity} = \frac{\lambda}{\rho c} \quad (14)$$

The solution of (13) is

$$\Delta T = \frac{Q}{8(\pi K t)^{3/2}} e^{\left\{ \left[(x-x')^2 + (y-y')^2 + (z-z')^2 \right] / 4Kt \right\}} \quad (15)$$

where Q is the strength of an instantaneous point source, i.e. the temperature to which the amount of heat liberated would raise a unit volume of the substance. The amount of heat liberated by the source is given by $Q\rho c$.

This interpretation of this solution is as the temperature in an infinite solid due to a quantity of heat $Q\rho c$ instantaneously generated at $t=0$ at the point (x', y', z') .

To solve (13) for a line source, integrate (15) along the z axis:

$$\Delta T = \frac{Q}{8(\pi K t)^{3/2}} \int_{-\infty}^{\infty} e^{\left\{ \left[(x-x')^2 + (y-y')^2 + (z-z')^2 \right] / 4Kt \right\}} dz \quad (16)$$

Therefore

$$\Delta T = \frac{Q}{4\pi K t} e^{\left\{ \left[(x-x')^2 + (y-y')^2 + (z-z')^2 \right] / 4Kt \right\}} \quad (17)$$

and the heat liberated per unit length of the line source is $Q\rho c$.

If heat is liberated at a rate $\phi(t)\rho c$ in unit time per unit length of a line parallel to the z axis and through the point (x', y') then the temperature at a time t after the supply of heat began is, from (17):

$$\Delta T(t) = \frac{1}{4\pi K} \int_0^t \phi(t') e^{-r^2/4K(t-t')} \frac{dt'}{t-t'} \quad (18)$$

where $r^2 = (x-x')^2 + (y-y')^2$

If $\phi(t) = q'$, a constant, then

$$\Delta T = \frac{q'}{4\pi K} \int_0^\infty \frac{e^{-u} du}{u} \quad (19)$$

$$\Delta T = \frac{-q'}{4\pi K} E_i\left(\frac{-r^2}{4Kt}\right) \quad (20)$$

where $-E_i(-x) = \int_x^\infty \frac{e^{-u}}{u} du$

For small x (large t or small r)

$$-E_i(-x) = \gamma + \ln(x) - x + \frac{1}{4}x^2 + O(x^3)$$

where $\gamma=0.5772\dots$ (Euler's constant)

Therefore

$$\Delta T = \frac{q'}{4\pi K} \ln\left(\frac{4Kt}{r^2}\right) - \frac{\gamma q'}{4\pi K} \quad (21)$$

This is the solution for temperature in the case of a solid heated by an infinitely thin wire carrying electric current.

Let q = quantity of heat produced per unit time per unit length of the wire.

Then

$$q = q'\rho c \quad (22)$$

Substituting this expression into (21) gives

$$\Delta T = \frac{q'}{4\pi\rho c K} \ln\left(\frac{4Kt}{r^2}\right) - \frac{\gamma q'}{4\pi\rho c K} \quad (23)$$

However, it is known that

$$K = \frac{\lambda}{\rho c} \quad (\text{from (14)})$$

Substituting into equation (23):

$$\Delta T = \frac{q'}{4\pi\lambda} \left[\ln\left(\frac{4Kt}{r^2}\right) - \gamma \right] \quad (24)$$

If $C = \exp(\gamma)$, then substituting gives

$$\Delta T = \frac{q'}{4\pi\lambda} \left[\ln\left(\frac{4Kt}{r^2}\right) - \ln(C) \right] \quad (25)$$

Therefore

$$\Delta T = \frac{q'}{4\pi\lambda} \ln\left(\frac{4Kt}{r^2 C}\right) \quad (26)$$

This result shows the linear dependence of temperature on $\ln(\text{time})$ for a thin wire within a solid or fluid material solely as a result of conduction processes.

Differentiating (26) with respect to $\ln(t)$ gives¹

$$\frac{d\Delta T}{d(\ln(t))} = \frac{q}{4\pi\lambda} \quad (27)$$

which yields an expression for λ , the thermal conductivity of the material under test:

$$\lambda = \frac{q/4\pi}{d\Delta T/d(\ln(t))} \quad (28)$$

For the purposes of this experiment, a variation on equation (28) is used:

$$\lambda = \frac{I^2 R}{4\pi L} \times \frac{dR}{d\Delta T} \times \frac{d(\ln(t))}{dR} \quad (29)$$

Equation (14) can then be used to determine the thermal diffusivity once the thermal conductivity is known.

3.2 THEORY OF ERRORS

While the experiment is being conducted, it must be realised that losses may occur and must be accounted for. The following are the main losses to be considered:

- i) Losses by conduction through the walls of the test chamber

For a free-standing wire, only the losses through the walls are of concern. As the walls are far from the wire (i.e. much greater than the diameter of the wire) these losses will not be significant.

- ii) Losses by radiation

The amount of heat lost in this manner is given by

$$Q = \sigma \left[\left(\frac{T_1}{100}\right)^4 - \left(\frac{T_2}{100}\right)^4 \right] F \quad (30)$$

where σ = Stefan-Boltzmann constant
 T_1, T_2 = temperatures of wire and fluid
 F = radiative surface area

As the wire used had a diameter of 50 μm , the surface area was very small (of the order of $3 \times 10^{-5} \text{ m}^2$) so these losses are also minimal.

iii) Losses by convection

The theory of measurement allows the effect of the onset of convection (namely, deviation of the gradient of dT versus $\ln(t)$ from a straight line) to be clearly seen in the output, and can thus be excluded from further consideration. In addition, convection may be delayed by constricting the volume of the test chamber.

iv) Eccentricity

If the axis of the wire does not coincide with the axis of the (cylindrical) test chamber, an eccentric correction is needed. For our purposes, the body of fluid can be treated as cylindrical.

3.3 ERRORS FROM PRINCIPAL EQUATION

Using Equation (30) the accuracy of the experimental apparatus can be determined:

$$\frac{\delta\lambda}{\lambda} = \frac{2\delta I}{I} + \frac{\delta R}{R} + \frac{\delta L}{L} + \frac{\delta d\Delta T}{d\Delta T} + \frac{\delta d\ln(t)}{d\ln(t)} \quad (31)$$

It is clear from this equation that accurate measurement of current (and the requirement of constant current) is essential. The resistance of the wire must also be accurately known, for not only is it involved in the above equation, but the apparatus may also be required to act as a thermometer to calibrate other experiments during the voyage to Titan and after landing.

Note that, given the equation defining resistance:

$$\frac{\delta R}{R} = \frac{\delta l}{l} - \frac{\delta a}{a} + \frac{\delta \rho}{\rho} \quad (32)$$

where l = length of wire
 a = cross-sectional area of wire
 ρ = resistivity

which can also be written⁵

$$\frac{\delta R}{R} = \frac{\delta l}{l} (1 + 2\nu) + \frac{\delta \rho}{\rho} \quad (33)$$

where ν = Poisson's ratio

indicates that, if it is possible to make an educated guess as to the amount of change in the length of the wire as a result of, for example, temperature change, shock of landing or the force of the fluid entering the testing chamber, it is possible to know how much error to allow for in the calculations. It

would appear to be reasonable to assume that resistivity will not change, given a small change in length (perhaps 1%). A 1% change in length would then give an approximately 1.6% change in resistance.

It is also crucial to know as accurately as possible the time interval (sampling rate) used.

The ideal experimental apparatus would consist on an infinitely long and infinitely thin wire surrounded by an infinitely thin layer of fluid. This would ensure that no temperature gradient forms in the fluid and hence no convection will occur. In practice, a wire of approximately 10cm in length and 50 μ m in diameter in a container approximately 12cm by 5cm by 3 cm was sufficient to delay the onset of convection for about 5 seconds.

4. EXPERIMENT

4.1 DETERMINATION OF THE TEMPERATURE COEFFICIENT OF RESISTANCE

After mounting the supplied 50 μ m wire in a plastic test box, a Keithley 195 digital multimeter was used to measure the change in resistance of the wire directly (to an accuracy of 10^{-3} Ω) while in the test chamber, hot water was allowed to cool or cold water allowed to warm up towards room temperature. The water was well stirred throughout to ensure a constant temperature, and readings of resistance were taken for every 0.5 $^{\circ}$ C change in temperature, as measured by a digital platinum resistance thermometer accurate to 0.1 $^{\circ}$ C.

Graphs were plotted of the results, Figures 4.1, 4.2 and 4.3 being those of cooling water and Figures 4.4, 4.5 and 4.6 those of warming water (i.e. from below room temperature). The results obtained showed a high degree of consistency and can be found in section 5.1. The original experimental data can be found in Appendix 1.

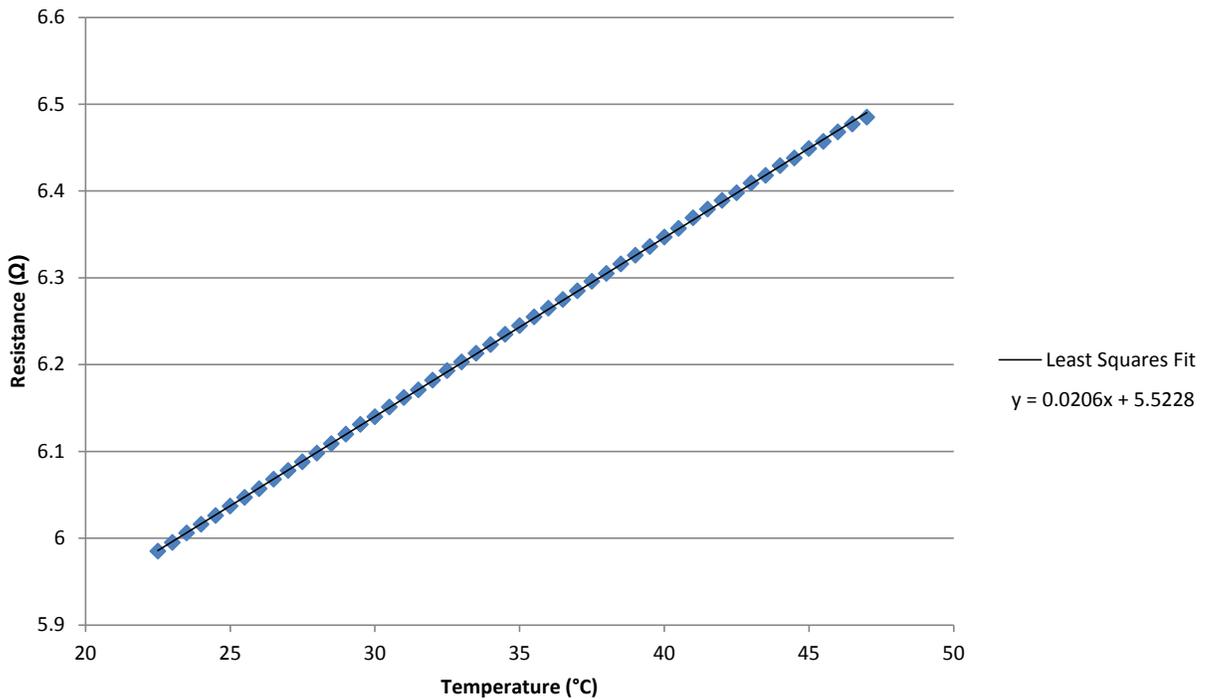


Figure 4.1 – First Cooling Run

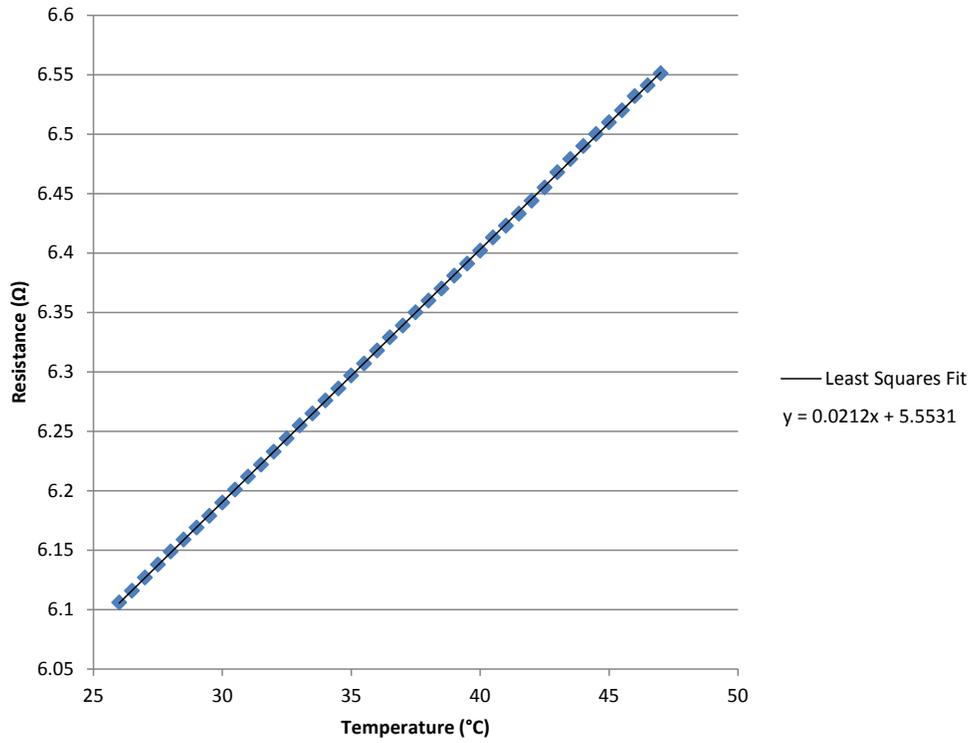


Figure 4.2 – Second Cooling Run

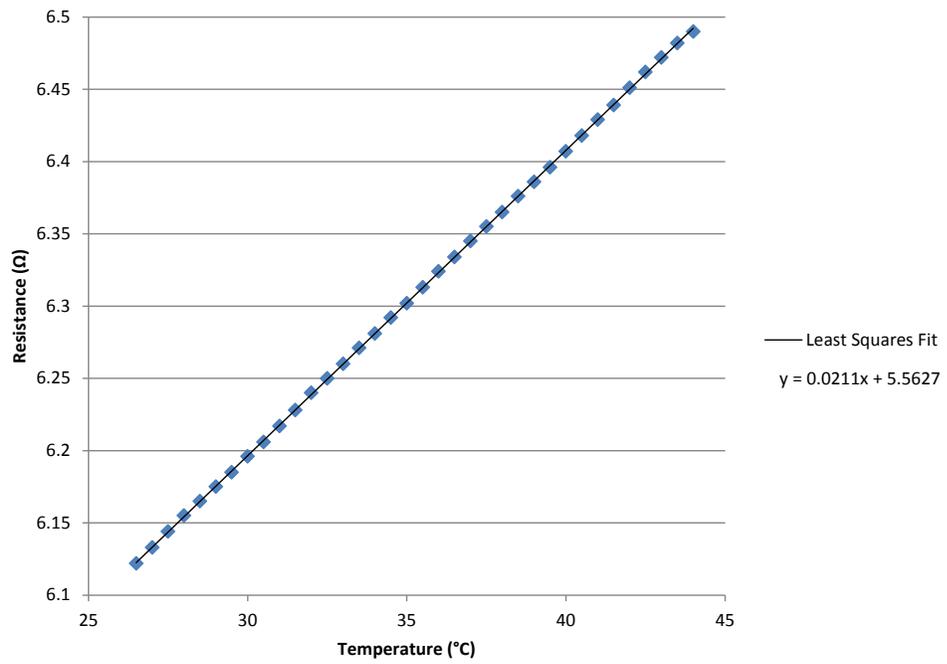


Figure 4.3 – Third Cooling Run

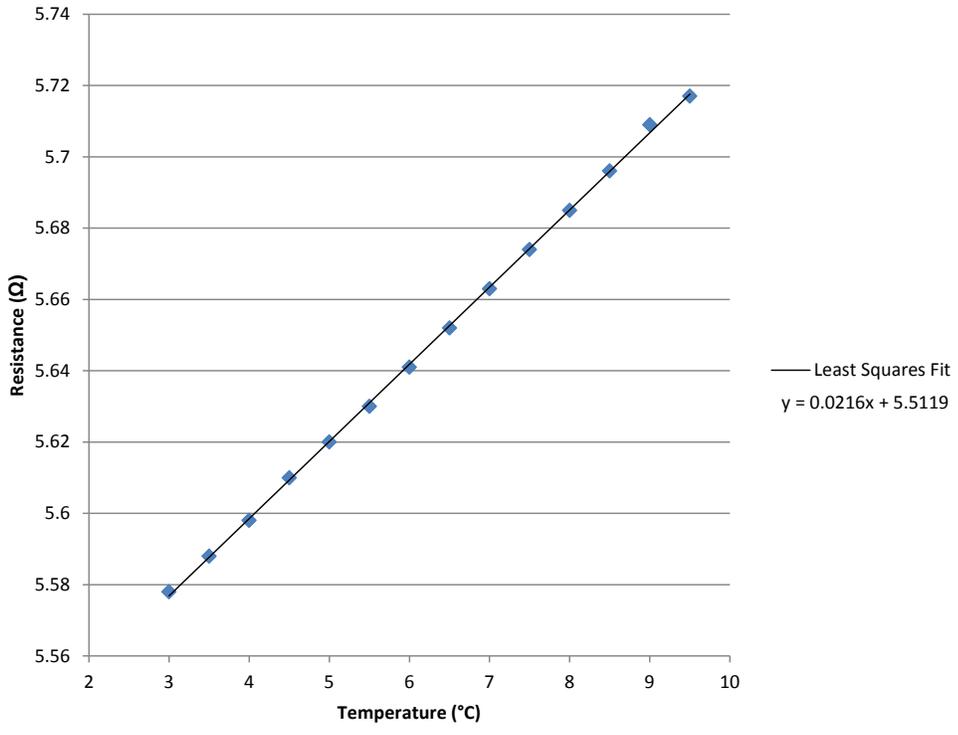


Figure 4.4 – First Warming Run

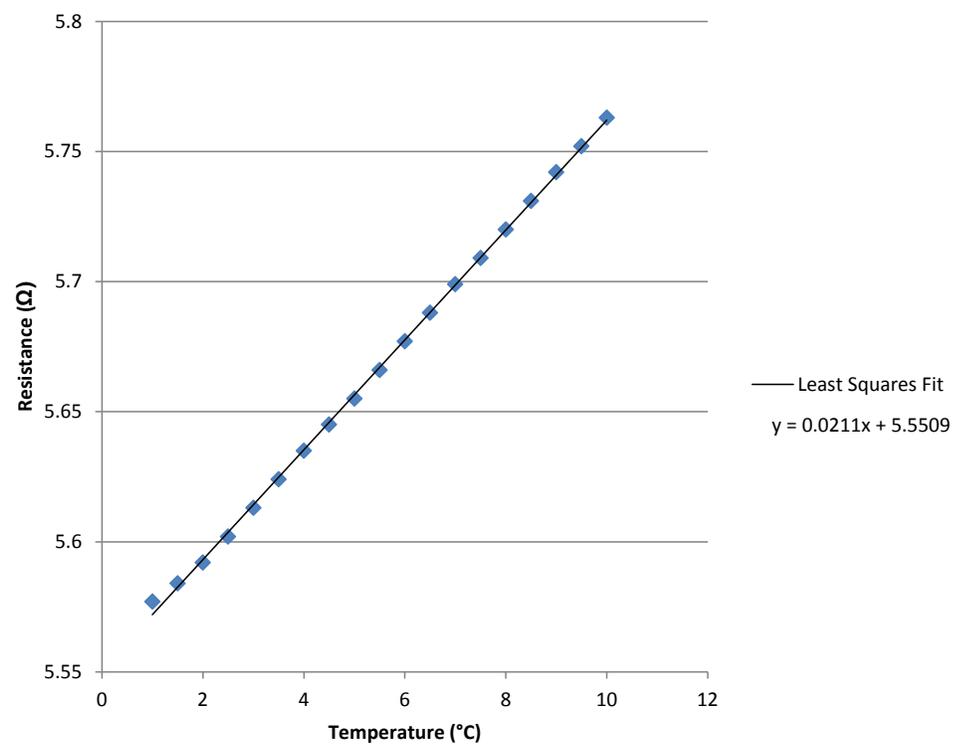


Figure 4.5 – Second Warming Run

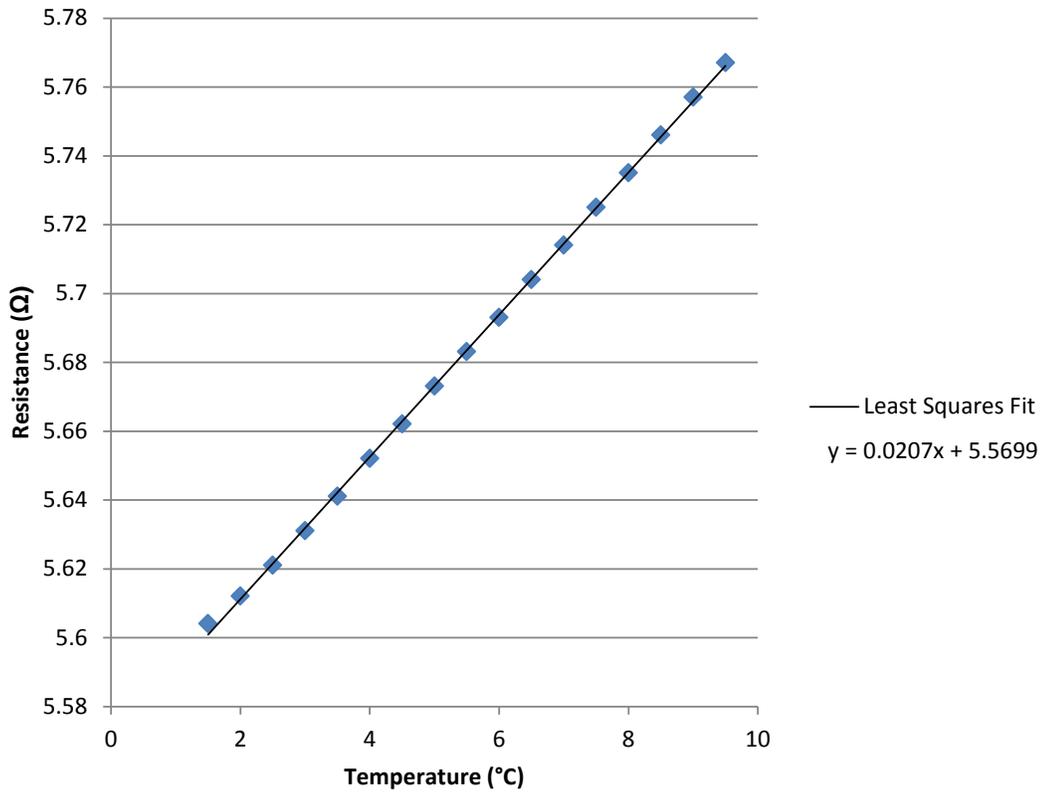


Figure 4.6 – Third Warming Run

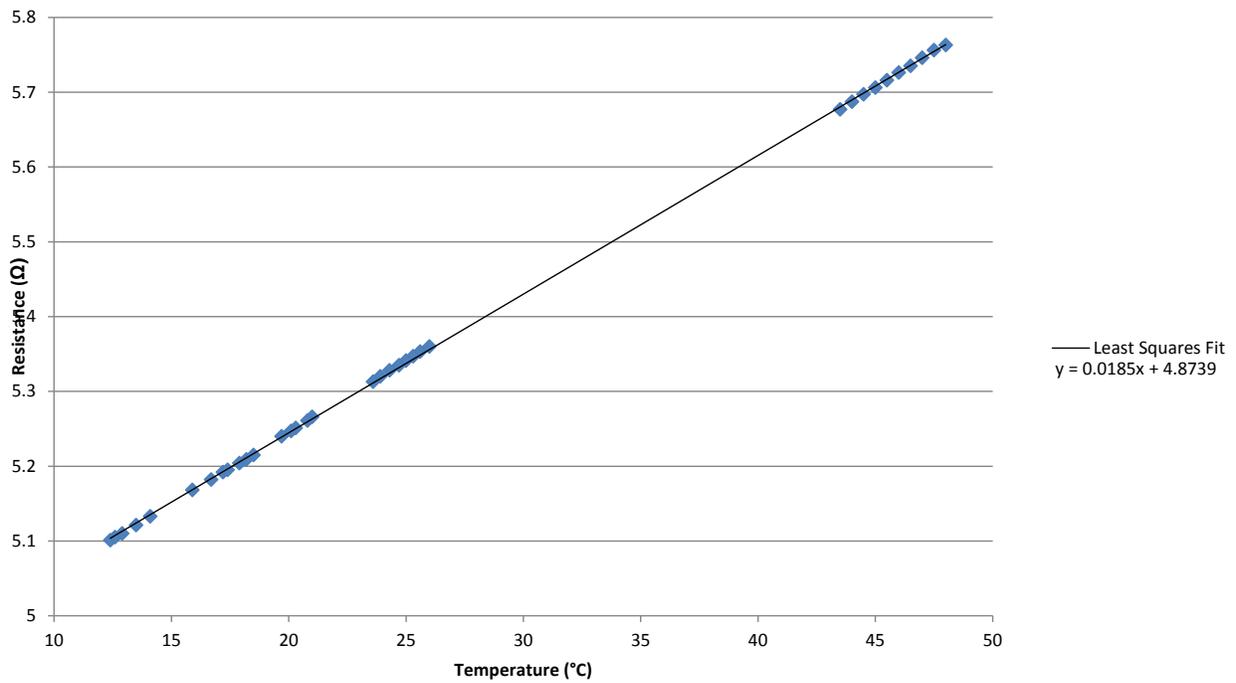


Figure 4.7 – Combined Warming and Cooling Runs with New Wire

Concerns were raised at this stage that the wire may have been soldered to the mounting in the test chamber incorrectly, resulting in a dry joint. When a new wire was fitted, the change of resistance with temperature was investigated – the result of this test is in Figure 4.7. This new wire was used to make the measurements of ΔR versus $\ln(\text{time})$ for the remainder of the project.

4.2 APPLICATION OF THE SUPPLIED TEST EQUIPMENT

It was necessary to write driver software for the test box, following the instructions given in reference 6. This was written in Turbo Pascal, this being the only compiled language available, and having sufficient graphics facilities for the needs of the project.

The test box itself caused a few problems – the 50-way “D”-type connection from the box to the PC30AT board was a female type, but male was required. A suitable connector was soon fitted. It was then found that the 4-way connections to the platinum wire had been connected to the wrong pins of its 9-way connector. Once discovered, this was soon rectified. After further problems with obtaining meaningful results, it was found that the polarity of the operational amplifier in the test box had been changed since the original instruction sheet had been written. It was then found necessary to alter the value of a resistance in the test box, as it had originally been intended to use a 25 μm diameter wire (resistance $\sim 25\Omega$) whereas a 50 μm diameter wire (resistance 5.2 Ω) was actually used. Once these initial teething troubles had been overcome, progress in making meaningful measurements was quickly made – raw data was obtained and a graph-plotting procedure was developed. This was further refined so that it was possible to exaggerate the vertical scale, so that once found, the section of interest could be examined more carefully. It was then possible to make approximate measurements from the monitor screen, printed screen dumps or by reference to the X and Y values of sample points in the section of interest. The vertical scale was altered to read first in volts, then in ohms, rather than the arbitrary units used by the test box. It was found that the wire’s container had to be kept as immobile in order to avoid forced convection due to vibration.

An attempt was made to sample in the $\ln(\text{time})$ domain, i.e. taking readings more frequently early on in the experiment and less frequently later on, as convection set in. It was decided that a constant sampling rate (10Hz) yielded sufficient sample values in the required area. Problems were found with obtaining exact timing of sampling – this was eventually overcome by utilising timers on the PC30AT board, the PC’s own timer proving inadequate for the task.

5. RESULTS

5.1 MEASUREMENTS OF VARIATION OF RESISTANCE WITH TEMPERATURE

Six sets of measurements were made, three for cooling and three for warming (to room temperature). As can be seen from the graphs (following pages) some sample points show a slight deviation from an otherwise straight line, usually at the “end” where measurements began. It is believed that these are not valid points, as it is unlikely that an even temperature would have been reached in the short time that those particular measurements were made.

The values of (dR/dT) derived from the graphs by the method of least squares are as follows:

Figure 4.1 \Rightarrow $0.0206 \Omega/^\circ\text{C}$
Figure 4.2 \Rightarrow $0.0212 \Omega/^\circ\text{C}$
Figure 4.3 \Rightarrow $0.0211 \Omega/^\circ\text{C}$
Figure 4.4 \Rightarrow $0.0216 \Omega/^\circ\text{C}$
Figure 4.5 \Rightarrow $0.0211 \Omega/^\circ\text{C}$
Figure 4.6 \Rightarrow $0.0207 \Omega/^\circ\text{C}$

The mean of these results is $0.0211 \Omega/^\circ\text{C}$

An estimate of the error involved is therefore

$$(1 - (0.0211/0.0206)) \times 100\% \approx 2.5\%$$

so

$$\delta(dR/dT)/(dR/dT) \approx 0.025$$

The new wire was similarly tested – the results may be seen in Figure 4.7. By the method of least squares, dR/dT was found to be $0.0185 \Omega/^\circ\text{C}$.

5.2 MEASUREMENTS OF RESISTANCE CHANGE AND LN(TIME)

A value of $d\Delta R/d\ln(t)$, from a curve fitted to the data is $0.00712 \Omega/\ln(s)$, however, this was obtained before accurate sample timing had been achieved. The screen dumps taken after timing correction gave values of $d\Delta R/d\ln(t)$ of $0.0075 \Omega/\ln(s)$.

From equation (29):

$$\lambda = \frac{I^2 R}{4\pi L} \times \frac{dR}{d\Delta T} \times \frac{d(\ln(t))}{dR}$$

where $dR/d\Delta T =$ gradient of graph from section 5.1 = $0.0185 \Omega/^\circ\text{C}$
 $I = 0.238 \text{ A}$
 $R = 5.3 \Omega$

$$L = 0.099 \text{ m}$$

$$d(\ln(t))/dR = \text{reciprocal of gradient of graph from section 5.2} = 200/1.5 = 133.33 \text{ ln(s)/}\Omega$$

gives a value for thermal conductivity λ for water at 20°C of

$$\lambda_{\text{water}} = 0.596 \text{ (to 3 decimal places)}$$

being a -0.3% difference from the expected value at 20°C (approximately 0.598).

Thermal diffusivity K is given by equation (14), thus given values of density and specific heat capacity of water at 20°C of

$$\rho = 998.21 \text{ kg/m}^3$$

$$c_p = 4181.8 \text{ J/kg } ^\circ\text{C}$$

then

$$K_{\text{water}} = 1.43 \times 10^{-7} \text{ m}^2/\text{s}$$

which is a 4% difference from the expected value at 20°C (approximately $1.49 \times 10^{-7} \text{ m}^2/\text{s}$).

5.2.1 Error Calculation

The total error in the experiment is given by the sum of the errors in $2I$, R , L , $d\Delta T$ and $d(\ln(t))$:

$$2\delta I/I = 0.02/0.25 = \pm 8 \times 10^{-3}$$

$$\delta R/R = 0.2/4.9 = \pm 0.04$$

$$\delta L/L = 0.002/0.099 = \pm 0.02$$

As $d\Delta T = \mu d\Delta R$,

$$\text{error in } d\Delta T = \text{error in } \mu + \text{error in } d\Delta R = \pm 0.05$$

$$\delta d(\ln(t))/d(\ln(t)) = 2/100 = \pm 0.02$$

$$\text{Total error budget} = \pm 14\%$$

The experimental results show an error of -0.3% for thermal conductivity and -4% for thermal diffusivity, so our calculation of $\pm 14\%$ puts our values near the true figure – the best which can be expected at this stage in testing. The timing must be done accurately to ensure accurate results.

6. CONCLUSION

The project was successful in its aim of verifying that the test box circuitry was capable of producing results which gave accurate values for the thermal properties of water, giving confidence that the method and proposed circuitry (suitably ruggedized) would be capable of similar performance on the surface of Titan. The circuit, the model for the one to be flown with the Surface Science Package, proved stable over a 20-bit range (a greater than expected accuracy). Whether to employ on-board processing to convert time data into logarithmic data, or to send the raw information directly to Earth, remains to be decided. The feasibility of the method has been established.

The graph plotting software must still be refined, perhaps using the refractometry experiment's software as a model. The method must still be tested with liquids other than water, for example toluene (around room temperature) and eventually liquid methane/ethane mixtures similar to those expected to be found on Titan.

Future experimentation should also consider the effect on accuracy of varying the sampling rate, for although the output of the test box is bandwidth limited to 10Hz, Nyquist's theorem implies that the maximum amount of accuracy may be obtained by reading this output at 20Hz or greater. It will at least be necessary to ensure a precise sampling rate – it has not so far proved possible to do this reliably in software. Ideally the current flow should also be computer-controlled. It is believed that errors in measurements encountered in this project are largely due to these timing inaccuracies.

7. REFERENCES

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Appendix 1 - Experimental Data

First Cooling Run

Temperature (°C)	Resistance (Ω)
47	6.485
46.5	6.477
46	6.468
45.5	6.457
45	6.449
44.5	6.438
44	6.429
43.5	6.418
43	6.409
42.5	6.398
42	6.389
41.5	6.379
41	6.369
40.5	6.357
40	6.347
39.5	6.336
39	6.326
38.5	6.316
38	6.305
37.5	6.296
37	6.285
36.5	6.275
36	6.265
35.5	6.255
35	6.245
34.5	6.235
34	6.223
33.5	6.213
33	6.203
32.5	6.193
32	6.182
31.5	6.171
31	6.162
30.5	6.151
30	6.14
29.5	6.131
29	6.12
28.5	6.109
28	6.098
27.5	6.088
27	6.078
26.5	6.068
26	6.057
25.5	6.047
25	6.037
24.5	6.026
24	6.016
23.5	6.006
23	5.995
22.5	5.985

Second Cooling Run

Temperature (°C)	Resistance (Ω)
47	6.551
46.5	6.541
46	6.532
45.5	6.52
45	6.51
44.5	6.5
44	6.49
43.5	6.479
43	6.468
42.5	6.455
42	6.444
41.5	6.433
41	6.423
40.5	6.413
40	6.402
39.5	6.391
39	6.381
38.5	6.37
38	6.36
37.5	6.35
37	6.339
36.5	6.329
36	6.318
35.5	6.307
35	6.297
34.5	6.286
34	6.276
33.5	6.265
33	6.255
32.5	6.244
32	6.233
31.5	6.222
31	6.212
30.5	6.201
30	6.19
29.5	6.179
29	6.169
28.5	6.159
28	6.149
27.5	6.138
27	6.127
26.5	6.116
26	6.106

Third Cooling Run

Temperature (°C)	Resistance (Ω)
44	6.49
43.5	6.482
43	6.472
42.5	6.462
42	6.451
41.5	6.439
41	6.429
40.5	6.418
40	6.407
39.5	6.396
39	6.386
38.5	6.376
38	6.365
37.5	6.355
37	6.345
36.5	6.334
36	6.324
35.5	6.313
35	6.302
34.5	6.292
34	6.281
33.5	6.271
33	6.26
32.5	6.25
32	6.24
31.5	6.228
31	6.217
30.5	6.206
30	6.196
29.5	6.185
29	6.175
28.5	6.165
28	6.155
27.5	6.144
27	6.133
26.5	6.122

First Warming Run

Temperature (°C)	Resistance (Ω)
3	5.578
3.5	5.588
4	5.598
4.5	5.61
5	5.62
5.5	5.63
6	5.641
6.5	5.652
7	5.663
7.5	5.674
8	5.685
8.5	5.696
9	5.709
9.5	5.717

Second Warming Run

Temperature (°C)	Resistance (Ω)
1	5.577
1.5	5.584
2	5.592
2.5	5.602
3	5.613
3.5	5.624
4	5.635
4.5	5.645
5	5.655
5.5	5.666
6	5.677
6.5	5.688
7	5.699
7.5	5.709
8	5.72
8.5	5.731
9	5.742
9.5	5.752
10	5.763

Third Warming Run

Temperature (°C)	Resistance (Ω)
1.5	5.604
2	5.612
2.5	5.621
3	5.631
3.5	5.641
4	5.652
4.5	5.662
5	5.673
5.5	5.683
6	5.693
6.5	5.704
7	5.714
7.5	5.725
8	5.735
8.5	5.746
9	5.757
9.5	5.767

Combined Warming and Cooling Runs with New Wire

Temperature (°C)	Resistance (Ω)
26	5.36
25.6	5.353
25.3	5.347
25	5.341
24.7	5.335
24.3	5.328
23.9	5.32
23.6	5.313
21	5.266
20.8	5.261
20.3	5.251
20.1	5.247
19.7	5.24
13.5	5.121
12.4	5.101
12.6	5.105
12.9	5.11
14.1	5.133
15.9	5.168
16.7	5.182
17.2	5.192
17.4	5.195
17.9	5.204
18.2	5.209
18.5	5.215
48	5.763
47.5	5.756
47	5.746
46.5	5.735
46	5.726
45.5	5.716
45	5.706
44.5	5.697
44	5.687
43.5	5.677